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New insights and advances on pyomelanin production: from microbial synthesis to applications

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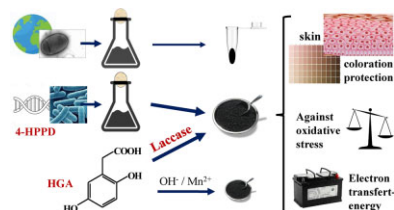
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Abstract: Pyomelanin is a brown-black phenolic polymer and results from the oxidation of homogentisic acid (HGA) in the L-tyrosine pathway. As part of the research for natural and active ingredients issued from realistic bioprocesses, this work re-evaluates the HGA pigment and makes an updated inventory of its syntheses, microbial pathways, and properties, with tracks and recent advances for its large-scale production. The mechanism of the HGA polymerization is also well documented. In alkaptonuria, pyomelanin formation leads to connective tissue damage and arthritis, most probably due to the ROS issued from HGA oxidation. While UV radiation on human melanin may generate degradation products, pyomelanin is not photodegradable, is hyperthermostable, and has other properties better than L-Dopa melanin. This review aims to raise awareness about the potential of this pigment for various applications, not only for skin coloring and protection but also for other cells, materials, and as a promising (semi)conductor for bioelectronics and energy.

Keywords: Pyomelanin, Polymerization, Hydroxylase, Laccases, Applications

Graphical abstract



Introduction

The true starting date is 1897 when a water-soluble, brown pigment-producing bacterium was isolated from a gummatous leg ulcer of a French cavalryman and was described as a “pyrocyanic bacillus” by Maxime Radais at the Faculty of Pharmacy in Paris. In 1902, began the ever-rebounding story of alkaptonuria (ALK), a rare, devastating, and osteoarticular disease, which aroused a lot of passion in the Great Ormond Street Hospital in London, and later all over the world. Paternity of the pigment origin undoubtedly belongs to La Du et al. (1958) and Zannoni et al. (1969), who demonstrated the involvement of the enzyme and the oxidation of homogentisic acid (HGA, 2,5-dihydroxyphenylacetic acid) that led to the ochronotic pigment found in connective tissue from patients with ALK. Thereafter, Yabuuchi & Ohshima (1972) confirmed these results and proposed the term “pyomelanin” for the pigment produced by *Pseudomonas aeruginosa*. The mutation of the gene responsible was revealed only in 1994 (Janocha et al., 1994). Just after, studies on *Aspergillus nidulans* have shown that the $\Delta hmgA$ mutant bears some analogy to results with

Aspergillus fumigatus, and the transfer of results to humans confirmed that polymers in ALK are associated with the defect of the HmgA protein (Fernández-Cañón & Peñalva, 1995; Schmalzer-Ripcke et al., 2009). Pyomelanin belongs to the heteroclite group of allomelanins encountered in all kingdoms in which melanins are produced from 1,8-DHN (1,8-dihydroxynaphtalene), 1,3,6,8-THN (1,3,6,8-tetrahydroxynaphtalene), catechols, GHB (γ -glutaminy-4-hydroxybenzene; see Supplementary Material), and HGA (see Fig. 1). To date, the HGA polymer still raises many questions about its functions and structure. The pigment has retained a poor image since its involvement in ALK disease and the low amounts generally produced by microorganisms. For these reasons, neither a production process nor an application has been developed until very recently. So what is the most appropriate method to produce pyomelanin, and how to identify microbial candidates?

HGA Pathways and Related Routes

Microorganisms are models of choice for HGA synthesis and production studies; some pathways have also been highlighted from

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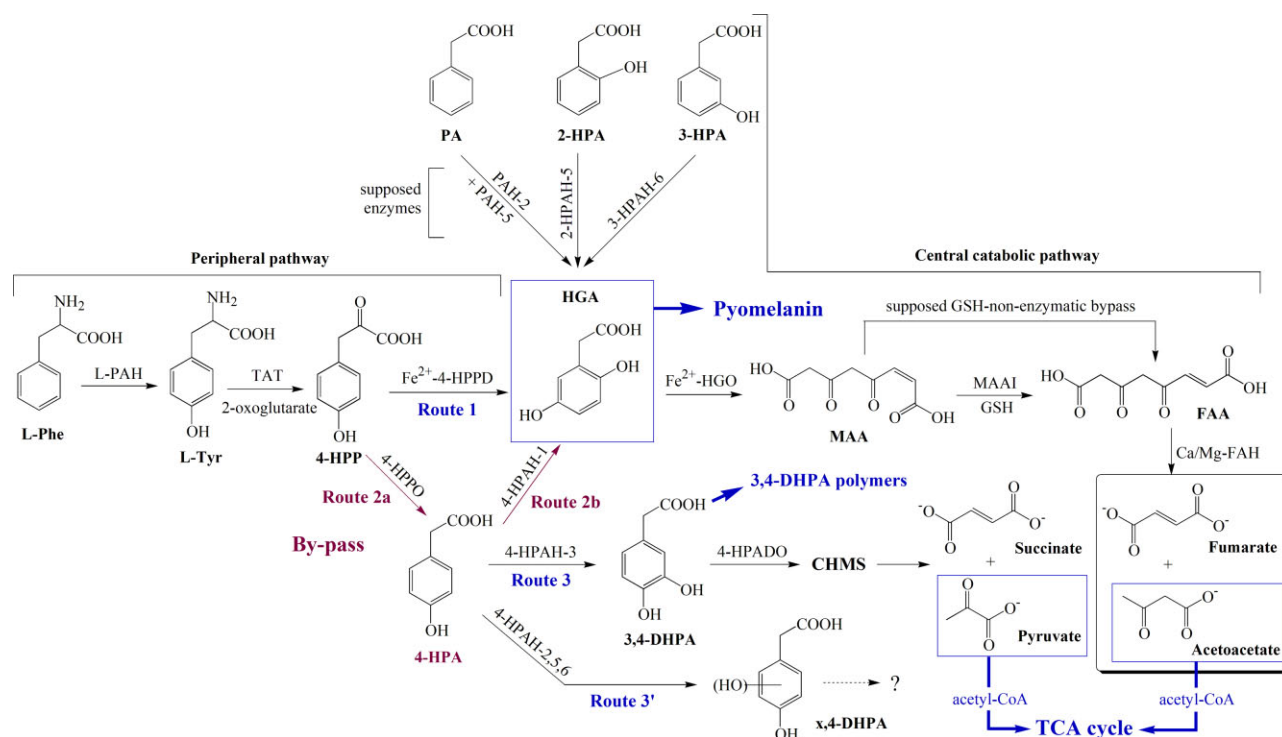


Fig. 1 Biosynthesis of pyomelanin through the homogentisate pathways is redefined as the HGA catabolon. This catabolon is involved in the degradation of L-Phe, L-Tyr, PA, and two hydroxy-PA derivatives into the common HGA and further toward the same route of catabolic convergence until fumarate and acetoacetate. Compounds: CHMS, 5-carboxymethyl-2-hydroxymuconic semialdehyde acid; 3,4-DHPA, 3,4-dihydroxyphenylacetic acid; FAA, fumarylacetoacetate; FAH, FAA hydrolase; GSH, glutathione; HGA, homogentisic acid; HGO, homogentisic acid 1,2-dioxygenase; x-HPA, x-hydroxyphenylacetic acid (x = 2, 3, or 4); 3,4-HPADO, 3,4-dihydroxyphenylacetate dioxygenase; 4-HPAH-y, 4-hydroxyphenylacetate hydroxylase (y = 1, 2, 3, 5, or 6); 4-HPP, 4-hydroxyphenylpyruvic acid; 4-HPPD, 4-hydroxyphenylpyruvate dioxygenase; 4-HPPO, 4-hydroxyphenylpyruvate oxidase; MAA, 4-maleylacetoacetate; MAAI, MAA isomerase; PA, phenylacetic acid; L-PAH, L-phenylalanine hydroxylase; PAH-z, phenylacetic acid hydroxylase (z = 2 or 5); L-Phe, L-phenylalanine; TAT, tyrosine transaminase; and L-Tyr, L-tyrosine.

studying mammalian cells with metabolic alterations to understand and potentially treat human pathologies. In the L-tyrosine (L-Tyr) degradation pathway, firstly L-phenylalanine (L-Phe) is converted to L-Tyr by the L-phenylalanine hydroxylase (L-PAH; EC 1.14.16.1), a non-heme iron-dependent enzyme. Then L-Tyr is converted to 4-hydroxyphenylpyruvate (4-HPP) by a tyrosine transaminase (TAT; EC 2.6.1.5) in the presence of 2-oxoglutarate as the cofactor. At this stage, two main distinct routes, 1 and 2 (Fig. 1), lead to HGA.

The Two Main HGA Routes

In route 1, the most widely reported, 4-HPP is converted to HGA by a non-hemic but Fe^{2+} -dependent enzyme, 4-hydroxyphenylpyruvate dioxygenase (4-HPPD, EC 1.13.11.27) (Raspail et al., 2011). Then HGA is transformed to 4-maleylacetoacetate (MAA) by a Fe^{2+} -dependent 1,2-dioxygenase (or HGA oxidase, HGO, EC 1.13.11.5), the rate-limiting step of L-Tyr catabolism, which opens the aromatic ring by an *ortho*-cleavage mechanism (C_1 - C_2 cleavage). MAA subsequently undergoes a *cis-trans* reversible isomerization catalyzed by a glutathione (GSH)-dependent maleylacetoacetate isomerase (MAAI, EC 5.2.1.2) and forming fumarylacetoacetate (FAA; Adachi et al., 1966). FAA is then hydrolyzed by a Ca/Mg-dependent FAA hydrolase (FAH, EC 3.7.1.2) in acetoacetate and fumarate, which will feed the Krebs cycle (Chapman & Dagley, 1962). In contrast, to human FAH, whose mutation causes hereditary tyrosinemia type I (TIT), MAAI failure does not produce disease. In addition, MAA can also be isomerized to FAA in the absence of MAAI, possibly

by a GSH-dependent non-enzymatic bypass mechanism through succinylacetoacetate (Fernández-Cañon et al., 2002). From MAA, a GSH-independent isomerase has been described occasionally in a *Bacillus* sp. strain and was shown to form maleate (instead of fumarate) and acetoacetate (Crawford, 1976). In the case of a mutation (like in ALK) or inactivation of the HGO activity, the HGA rises in excess and is excreted from the cell, where it is oxidized rapidly in the presence of O_2 to form pyomelanin.

In route 2 (2a and 2b), the least studied, 4-HPP is converted to 4-hydroxyphenylacetate (4-HPA) by a 4-HPP oxidase (4-HPPO, EC 1.2.3.13) (Blakley, 1977). At this stage, a NAD(P)H-dependent key enzyme, 4-HPA-1-hydroxylase (4-HPAH-1, EC 1.14.13.18) hydroxylates the ring at the C_1 position. HGA is formed after transposition, named the “NIH shift” for the National Institute of Health, the research unit of origin; the shift concerns the acetic group $-\text{CH}_2-\text{COOH}$, which moves to carbon C_2 (Hareland et al., 1975; see the mechanism of hydroxylation in Supplementary Material). Then, the HGA ring is opened by the HGO enzyme leading to the synthesis of the end products, similarly to route 1. C_1 -hydroxylation and HGA degradation were demonstrated in *Delftia acidovorans* (formerly *Pseudomonas* or *Comamonas acidovorans*; Hareland et al., 1975), *Bacillus* sp. and *Moraxella* sp. (Crawford, 1976), *Vibrio cholera* (Kotob et al., 1995), *Shewanella colwelliana* (Coon et al., 1994), *Pseudomonas putida* (Arias-Barrau et al., 2004), *Xanthobacter* sp. (van den Tweel et al., 1986), *Flavobacterium* sp. (van den Tweel et al., 1988), *Azoarcus evansii* (Mohamed et al., 2002), *Cryptococcus neoformans* (Frases et al., 2007), *Halomonas olivaria* (Amouric et al., 2014; Liebgott et al., 2008), and many others. C_1 -hydroxylation is much less encountered than C_3 -hydroxylation,

which is generally carried out by a 4-HPAH-3 enzyme in most microorganisms by forming 3,4-dihydroxyphenylacetate (3,4-DHPA; route 3, Fig. 1). This *ortho*-diphenol polymerizes much less rapidly than HGA in bacteria, such as *Klebsiella pneumoniae* (Gibello et al., 1995), *Serratia marcescens* (Trias et al., 1989), but not in *Halomonas* sp. HTB24 because of the concomitant presence of hydroxytyrosol (or 3,4-dihydroxyphenylethanol), another *ortho*-diphenol, and a powerful antioxidant (Liebgott et al., 2007, 2009).

Other Routes and the Homogentisic Catabolon Concept

More rarely, hydroxylation on carbon other than C₁ by other hydroxylases (HPAH, Fig. 1) had also been identified from HPA isomers other than 4-HPA. They involve more specific transport systems, usually generate minor dihydroxy compounds, and may reveal the multipotential capacity of some microorganisms in the degradation of xenobiotics. HGA was synthesized at low quantities from 2-HPA (C₅ hydroxylation) in *Pseudomonas fluorescens* (Baggi et al., 1983) and *Rhodococcus erythropolis* (Suemori et al., 1996) and 3-HPA (C₆) in *Trichosporon cutaneum* (Anderson & Dagley, 1980), *P. putida* U (Arias-Barrau et al., 2004, 2005), *R. erythropolis* (Suemori et al., 1996), *A. nidulans* (Ferrer-Sevillano & Fernández-Cañón, 2007), and *Burkholderia xenovorans* LB400 (Mendez et al., 2011). In addition, the random mutant of *Beauveria bassiana* fungus strain L6577 has been shown to synthesize HGA from phenylacetic acid (PA), 2-HPA, 3-HPA, and 3-methoxy-PA, but the yield of HGA has not been reported (Staudenmaier et al., 1999). Surprisingly, L6577 provides a conversion of 4-chloro-PA to 3 g L⁻¹ of 4-chloro-HGA, a photographic developer and a very useful compound for the preparation of pharmaceuticals.

Aromatics can be degraded by a group of enzymes that belong to a complex functional catabolic unit, the catabolon that integrates different routes of the upper pathways. These routes generally catalyze the transformation of structurally related compounds into a common catabolite, in this case, HGA. Such a concept is novelty, defined here as the HGA catabolon encompasses all the routes involved in the transformation of L-Phe, L-Tyr, 4-HPP, PA, 2-HPA, 3-HPA, and 4-HPA into HGA, which is the common intermediate (Fig. 1). HGA is subsequently catabolized through a route of convergence, the HGA catabolon core, into general metabolites.

Pathways Coexistence With Pyomelanin

At this date, routes 1 and 2 (Fig. 1) have never been reported to coexist in the same microorganism or species, most probably because the investigations have focused mainly on the 4-HPPD rather than the 4-HPAH-1 enzyme. The same observation has been made for routes 2 (HGA bypass) and 3 (3,4-DHPA formation). In contrast, routes 1 and 3 (through 2a) coexist in bacteria of the same species that possess an extensive advanced cell factory due to their metabolic plasticity and better adaptation to different environments. In the degradation of L-Tyr and biogenic aromatic amines, this is especially the case for *P. aeruginosa* (Cuskey & Olsen, 1988; Rodríguez-Rojas et al., 2009) and *P. putida* (Arcos et al., 2010; Paliwal et al., 2014). Secondly, in a few cases, the coexistence of several pathways and melanin types has also been encountered in the same organism or species. In bacteria and yeasts, two routes occurred, one tyrosinase-dependent involved in 3,4-dihydroxyphenylalanine (L-Dopa) synthesis and another tyrosinase-independent pathway, however, individually highlighted by the degradation of substrates added separately in the culture. As an illustration, pyomelanin was formed from L-Tyr (Carreira et al., 2001; Kurian & Bhat, 2018), and L-Dopa melanin

from L-Tyr (Ganesh Kumar et al., 2013) or L-Dopa (Apte et al., 2013), in *Pseudomonas stutzeri* and *Yarrowia lipolytica*. Besides that, the case of fungi remains particularly confusing. Starting from acetyl-CoA or malonyl-CoA, they generally synthesize melanin from 1,8-DHN by a laccase (Sapmak et al., 2015; Upadhyay et al., 2013). Alternatively, a few fungi like *C. neoformans* known to form melanin from exogenous substrates, produce pigmentation from L-Dopa, catechol, and other catecholamine precursors (epinephrine, norepinephrine) by laccase(s) (García-Rivera et al., 2005; Williamson, 1994), as well as pyomelanin from HGA (Frasés et al., 2007). And if L-Tyr is the precursor, it is first converted to L-Dopa and dopaquinone by tyrosinases, then to L-Dopa-melanin. As another case, from L-Tyr, the pathogenic fungi *A. fumigatus* produces DHN-melanin as a virulence factor and pyomelanin as a defense function (Heinekamp et al., 2013; Schmalder-Ripcke et al., 2009).

Identify the HGA Pathway First Approach by the Relevant Metabolites

Identifying the routes through the degradative intermediates remains possible if growth is slowed down by depleting the C and N sources of the medium. The strain preferentially utilizes the aromatic compounds over glucose, their degradation can be followed along with the growth by RP-HPLC-DAD (reverse-phase-HPLC-diode array detector), RP-HPLC-MS (mass spectrometry coupling in electrospray ionization [ESI] mode), and/or GC-MS (gas-liquid chromatography-mass spectrometry in electron-impact ionization [EI] mode) of the derivatized metabolites (Liebgott et al., 2007–2009). Accurate identification is achieved by their respective retention time, UV-visible spectra, and *m/z* mass data, over all standards available and spectra from the databank (NIST or Wiley). Cultures must be induced by appropriate substrates assayed separately at concentrations not exceeding 1–5 mM in the medium: (i) L-Tyr or even tyrosol, a phenolic strongly accumulated in olive-wastewaters, for the global identification of the intermediates at different times, (ii) 4-HPP (route 1 or 2), (iii) 4-HPA (route 2b, 3, or 3'), and (iv) HPA isomers (HPA, 2-HPA, and 3-HPA) for other HGA routes and through the respective hydroxylases involvement (Liebgott et al., 2007–2009; Lorquin et al., 2021; Turick et al., 2009) (Fig. 1). HGA and pyomelanin were shown to increase in the presence of an excess of 4-HPA and 4-tyrosol during the growth of *Halomonas titanicae* and *Halomonas Olivaria*, respectively, these two bacteria being unable to grow in the presence of L-Tyr (Liebgott et al., 2008; Lorquin et al., 2021). Identical strategies had also been used in fungi and yeasts (Almeida-Paes et al., 2017; Schmalder-Ripcke et al., 2009). By HPLC-DAD analyses of the culture supernatant, 4-HPA (λ_{\max} 274 nm), 4-HPP (λ_{\max} 295_{weak}, 311 nm), and HGA (λ_{\max} 290 nm) must be searched first and confirmed by mass spectrometry (ESI or EI mode) to propose routes 1 and/or 2. Metabolites directly issued from the degradation of HGA might also be identified by ion-exchange HPLC-DAD, MAA, and FAA (λ_{\max} 315–330 nm each, depending on the pH) (Bergeron et al., 2001; Chapman & Dagley, 1962). The increase of acetoacetate and fumarate amounts evaluated by either ion exchange HPLC-refractive index detection or GC-MS analyses will confirm the L-Tyr catabolism via HGA routes.

By Inhibition of the Key Enzymes

Inhibitors assayed separately are essential to decide between routes 1 or 2 (Fig. 1). As a global approach, better than *p*-chloro- and α -methylphenylalanine (Kelly & Johnson, 1978), a synthetic inhibitor acting as a small binder of the L-PAH enzyme, named

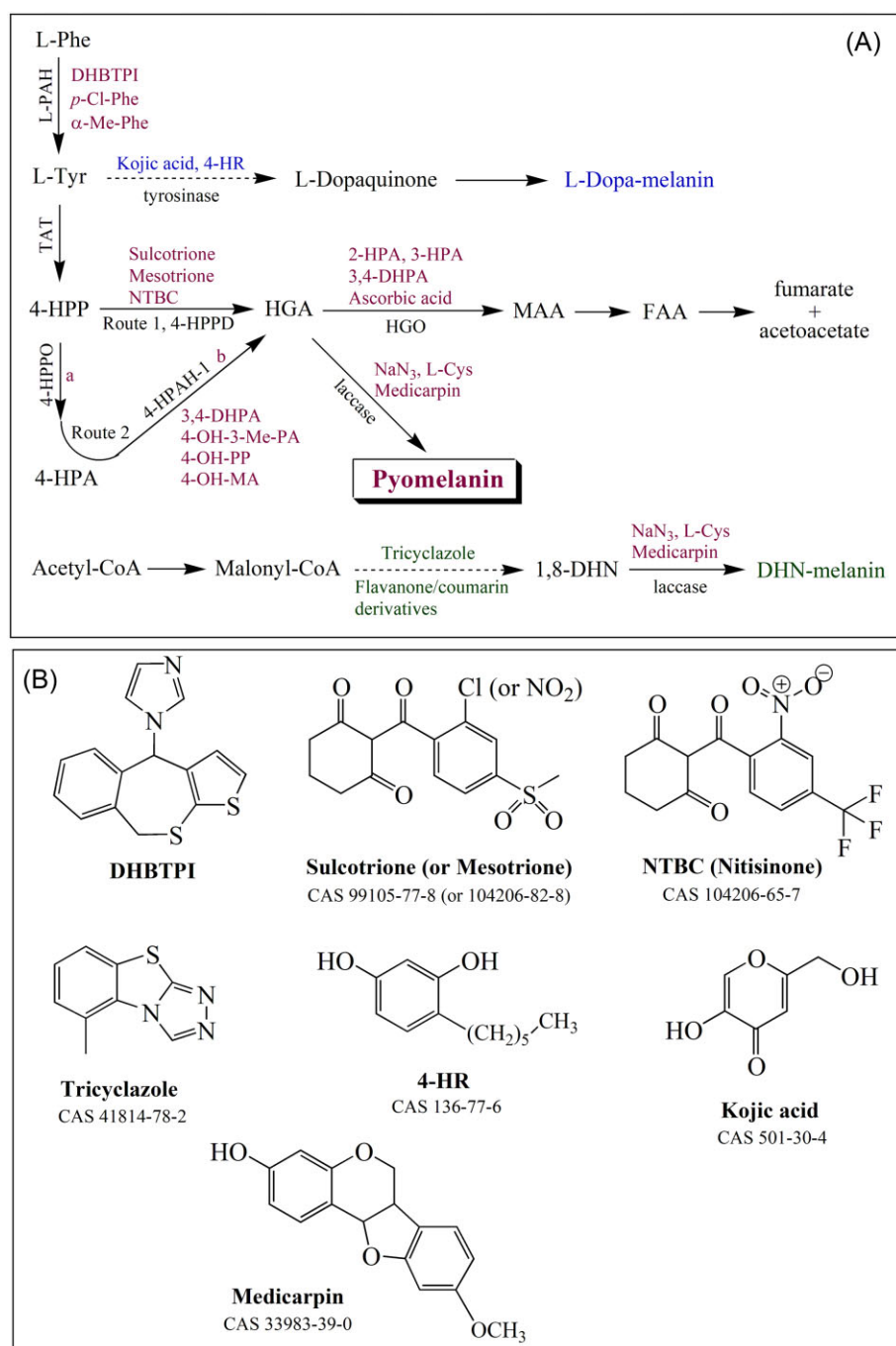


Fig. 2 Inhibitors of pyomelanin synthesis and other inhibitors to differentiate pyomelanin from other melanin. (A) Inhibitors of pyomelanin synthesis are represented in red, compared to those of L-Dopa (blue) and DHN-melanin (green). Reactions are detailed in Fig. 1. (B) Chemical structure of the main inhibitors. Abbreviations: L-Cys, L-cysteine; DHBTPI, 1-(4,9-dihydrobenzo[e]thieno[2,3-b]thiepin-4-yl)-1H-imidazole; 4-HR, 4-hexylresorcinol; kojic acid, 5-hydroxy-2-(hydroxymethyl)-4H-pyran-4-one; MA, mandelic acid; medicarpin, 3-hydroxy-9-methoxypterocarpan; mesotrione, 2-[4-(methylsulfonyl)-2-nitrobenzoyl]cyclohexane-1,3-dione; NaN₃, sodium azide; NTBC (nitisinone), 2-[2-nitro-4-(trifluoromethyl)benzoyl]cyclohexane-1,3-dione; PA, phenylacetic acid; PP, phenylpropionic acid; sulcotrione, 2-[4-(methylsulfonyl)-2-chlorobenzoyl]cyclohexane-1,3-dione; and tricyclazole, 5-methyl-1,2,4-triazolo(3,4-b)benzothiazole.

DHBTPI (IC₅₀ 3.8 μM; Aubi et al., 2015) must be assayed (see Fig. 2). DHBTPI does not affect the growth of *Legionella pneumophila* but abolishes the synthesis of pyomelanin. Experiments should be further organized around two ways: (A) inhibition of 4-HPPD (route 1) and 4-HPAH-1 (route 2b) enzymes; and (B) inhibition of HGO enzyme to promote HGA accumulation.

Point A. 4-HPPD (EC 1.13.11.27), the main described enzyme, is a Fe²⁺-dependent non-heme protein that belongs to the class

of α-keto acid-dependent oxygenases with different roles in prokaryotes, plants, and animals. The enzyme has only two substrates, 4-HPP as the source of α-keto acid and molecular oxygen, and differs from most α-keto acid-dependent oxygenases that normally require α-ketoglutarate. The mechanism was suggested by Borowski et al. (2004), followed by works by Raspail et al. (2011) and Wójcik et al. (2014). In the last four decades and for weed control, agrochemical groups have identified new herbicides as

potential 4-HPPD inhibitors that belong to pyrazoles, triketones, and isoxazoles (Ndikuryayo et al., 2017; van Almsick, 2009). Furthermore, these developments have made it possible to inhibit microbial 4-HPPDs successfully. Some inhibitors are derivatives of the natural phytotoxin leptospermone, and among them, nitisinone (or NTBC, see Fig. 2) was authorized in 2002 (USA) and 2005 (EU) to treat ALK, TIT, and hawkinsinuria diseases (Arnoux et al., 2015; McKiernan, 2006; Santucci et al., 2017). Finally, three inhibitors belonging to the triketone class, sulcotrione, mesotrione, and NTBC are commonly used to inhibit route 1 and rule on the involvement of 4-HPPD (Fig. 2). The 4-HPAH-1 enzyme (EC 1.14.13.18; route 2b) studies were initiated by those on genistic acid (2,5-dihydroxybenzoic acid; Crawford, 1975; Crawford et al., 1975). The substrates are 4-HPA, NAD(P)H, and O₂, and the activity needs FAD as the cofactor and Mg²⁺, the mechanism was never detailed afterward, but it was suggested by analogy to that of 4-HPAH-3 (Arias-Barrau et al., 2004; Liebgott et al., 2009; Prieto & Garcia, 1994). If Suemori et al. (1996) then Zink et al. (2012) purified the enzyme from *R. erythropolis* and *D. acidovorans*, respectively, ultimately, inhibition of 4-HPAH-1 was only reported by Hareland et al. (1975) and assumed mainly by 3,4-DHPA, 4-hydroxy-3-methyl-phenylacetic acid, 4-hydroxyphenylpropionic acid, and 4-hydroxymandelic acid (Fig. 2).

Point B. To accumulate pyomelanin, the inhibition or inactivation of the HGO activity would be a solution. The crystal structure of the human enzyme was first proposed by Titus et al. (2000) and the mechanism suggested by Borowski et al. (2005). Although HGO inhibition has not been studied extensively, the enzyme from various sources was successfully inhibited or inactivated by HPA analogs and could serve for other microbial HGO inhibition studies (Fig. 2 and Supplementary Material). Instead of using HPA analogs that might be incorporated and alter the structure of the final polymer, the absence of iron in microbial cultures is preferred to inactivate HGO (Liebgott et al., 2008).

Doses range of the main inhibitors used. Kojic acid (10–1000 μg mL⁻¹), tricyclazole (10–100 μg mL⁻¹), sulcotrione or mesotrione (50–100 μg mL⁻¹; Fang et al., 2010; Tan et al., 2019), DHBTP1 (10 μM; Aubi et al., 2015), 2- and 3-HPA (1–2 mM; Sugumaran & Vaidyanathan, 1978), 4-HR (10–100 μM; Dawley & Flurkey, 1993), medicarpin (0.1–0.5 mM; deduced from Martinez-Sotres et al., 2012, 2015), NaN₃ (0.01–1 mM), L-Cys (0.1–1 mM; Couto & Toca, 2006), and NTBC (0.1–1 mM; Ketelboeter et al., 2014).

Differentiate Pyomelanin From Other Pigments

In patients' urine and as a suitable method for the diagnosis of ALK, the addition of NaOH plus hypochlorite accelerates the polymerization of HGA, showing two additional peaks at 406 and 430 nm in the visible spectrum and suggested to be HGA conjugated-bilirubin compounds (Tokuhara et al., 2018). In microorganisms, the possible coexistence of pathways and melanin might make the identification of pyomelanin among allomelanins and L-Dopa-melanin more complex. Inhibitors must help to differentiate the melanin types and pathways.

For an initial assessment, tricyclazole, an inhibitor of 1,8-DHN melanin synthesis (prevent the reduction of 1,3,8-THN and 1,3,6,8-THN), kojic acid, arbutin, and 4-hexylresorcinol (see Fig. 2), three tyrosinase inhibitors (L-Dopa pathway), are the most currently assayed in cultures, pyomelanin would not be affected by these treatments. Other flavanone and coumarin-derived compounds that inhibit DHN-melanin synthesis (Ganesh Kumar et al., 2011) and other tyrosinase inhibitors (reviewed in Chang, 2009) could be used as well. Secondly, the laccase inhibitors, Na-azide,

L-cysteine, and the most recent medicarpin, may be assayed to prevent the HGA and/or DHN oxidation-polymerization, HGA and/or DHN would normally accumulate (Fig. 2). By such inhibition treatment, the excluded melanin must further be identified by FTIR (Fourier Transform InfraRed spectroscopy), ¹³C solid-state NMR (or ¹³C cross-polarization magic angle spinning NMR), alkaline-H₂O₂ (or permanganate) hydrolysis, and/or pyrolysis-GC-MS methods.

Some chemical features can also help to elucidate the melanin structures in a mixture. The very first difference is that eumelanin contains N, pheomelanin N and S, while allomelanins contain neither. Since pyomelanin and Dopa-melanin are dark-brown pigments, DHN-melanin is green-black, and GHB-melanin is reddish-brown. Moreover, the configuration of quinone or quinonimine residues is different from one class of melanin to another. It is *ortho* in L-Dopa-melanin, *para* in pyomelanin and GHB-polymers, and *meta* in 1,8-DHN melanins. Unfortunately, there are analytical brakes for the identification of pyomelanin, in contrast to L-Dopa melanin, for which compiled data of MALDI-ToF (matrix-assisted laser desorption/ionization-time of flight) mass spectrometry, NMR, and pyrolysis-GC-MS exist and complement degradative techniques (Pralea et al., 2019). Indeed, alkaline-H₂O₂ oxidation of L-Dopa melanin led to the identification of four possible degradation products, pyrrole-2,3-dicarboxylic acid, pyrrole-2,5,5-tricarboxylic acid, thiazole-4,5-dicarboxylic acid, and thiazole-2,3,5-tricarboxylic acid (d'Ischia et al., 2013; Wakamatsu et al., 2003). Such hydrolysis treatments failed for pyomelanin structure determination (Lorquin et al., 2021). A few researchers also tried to characterize pyomelanin by MALDI-ToF and ESI-MS-MS (tandem) mass spectrometry but gave inaccurate molecular weight and unknown fragments (Roberts et al., 2015; Singh et al., 2018). Consequently, priority must be given to the development of a specific matrix and ionization mode for the HGA polymer sequencing. Pyrolysis-GC-MS coupling methods also failed from pyomelanin of *Penicillium chrysogenum* (Vasanthakumar et al., 2015), and again from our HGA pigments (Lorquin et al., 2021), giving mainly 4-methoxybenzene acetic acid and 4-methoxybenzene propanoic acid, but no HGA. At least pyomelanin is negatively charged, resistant to acids, bleaching appears but not always when subjected to oxidizing agents, exhibits a positive reaction with FeCl₃ and Na-dithionite, and is insoluble in most solvents. All that still tends the differentiation between melanin difficult. Its insolubility in organic solvents, however, may easier its purification from media that do not contain other melanin.

In Vitro Polymerization and Transformations of HGA

For a long time, pyomelanin had been thought to be the result of the aerobic autoxidation of HGA from neutral to alkaline pH (Zannoni et al., 1969), followed by self-assembly into polymers. The oxidation is enhanced especially by Mn²⁺ that doubles the reaction yield, accelerated by SOD and Mn-pyrophosphate, and inhibited by NADH, GSH, and ascorbic acid (Coon et al., 1994; Martin & Batkoff, 1987). Formation of the initial oxidized product, 1,4-benzoquinone acetic acid (BQA), was proportional to the HGA concentration, optimal at 37°C, inferred by decreased UV-absorption at 290 nm (HGA) and increase at 250 nm (appearance of BQA). Toxic ROS, such as O₂⁻ HO[•], and quinone radicals are rapidly formed and play a significant role in the etiology of ALK arthritis. While pyomelanin issued from the abiotic HGA autoxidation has been commonly reported (Ruzafa et al., 1995;

Schmaler-Ripcke et al., 2009), the reaction had never been optimized until recently when an (HGA)/(Mn²⁺) ratio of 20 has been fixed for an optimal pyomelanin yield. By applying this ratio, an online process yielded 0.317 g of pyomelanin per g of 2,5-dimethoxyphenylacetic acid (2,5-DMPA, the starting compound; Table 1 and Fig. 3), a method that is easy to implement but limited to uses on a test lab scale. At alkaline pH, a chemical substitution at the C₄ position of the BQA ring by added primary and secondary amines, had been described by Stoner & Blivaiss (1967) and might occur in biological systems where amino acids and proteins are present. By changing the pH from 7 to 4–5, BQA previously formed from HGA was shown to be chemically converted into 2,5-dihydroxybenzaldehyde (or gentisaldehyde) in the presence of Mn²⁺ or Cu²⁺ (Liebgott et al., 2008) (Fig. 3), whereas from HGA no reaction was observed at pH 6.5 (La Du & Zannoni, 1963). Gentisaldehyde is the resulting product of an oxidative decarboxylation reaction, a mechanism like that observed with 3,4-DHPA (Mefford et al., 1996).

In Biological Systems: The Role of Laccases

Toxic ROS formed during the oxidation of HGA can deplete systemic or local antioxidants, leaving the remaining oxidants to increasingly react with essential proteins, DNA, and other macromolecules to cause or contribute to tissue toxicities (Hegedus & Nayak, 1994). In biological systems, like Mn²⁺, lactic acid, L-asparagine, and glycine promote coloration by contributing to the alkalization of the medium and the autoxidation of HGA (Carreira et al., 2001). Another harmful effect concerns the cellular HGA oxidation at an alkaline pH in the presence of copper. Indeed, H₂O₂ can be chemically generated during Cu(II)-catalyzed HGA autoxidation and reacts with Cu(I) to form a Cu(I)-peroxide complex, this complex notably could cause oxidative DNA damage at thymine and cytosine residues (Hiraku et al., 1998). Is the chemical autoxidation of HGA the sole cause of its polymerization? While they exhibit similitudes with FTIR data (Schmaler-Ripcke et al., 2009) showing the best spectral resolution for lower size polymers, synthetic and microbial pyomelanin are structurally different (David et al., 1996; Lorquin et al., 2021; Zheng et al., 2013). Since polyphenol oxidases (PPOs) are copper enzymes, living cells also contain non-copper enzymatic systems able to catalyze phenol hydroxylation, such as cytochrome P450, 4-HPA, and 4-HPP hydroxylases, which may be considered as PPOs and which do not play a central role in physiological melanin formation. At pH 6.5, horseradish peroxidase could not polymerize HGA, while in the presence of GSH, the synthesis of 3,6-diglutathione-HGA occurred (La Du & Zannoni, 1963). Later, BQA had been shown to serve as a substrate for a partially purified copper-dependent oxidase from albino guinea pig skin (Zannoni et al., 1969). Logically suspected, tyrosinases were characterized in many melanin-producing microorganisms but have never been involved in HGA oxidation. Several authors hypothesized the action of laccases (EC 1.10.3.2) to polymerize *p*-dihydroxylated monophenols, as suggested in the bacteria *Vibrio cholerae* (Ruzafa et al., 1995) and *Alcaligenes eutrophus* (now *Cupriavidus necator*; David et al., 1996). In the wild-type *C. neoformans* B3501 strain, pyomelanin production was confirmed to be laccase-dependent by using a deleted construction (2E-TU Lac mutant), the enzyme expression is repressed by glucose that affects pigment induction (Frases et al., 2007). Pyomelanin-forming bacteria generally grow at pH 6–7, while the HGA autoxidation is optimal at pH 8–9, one more element in favor of the laccase(s)

action. At least, a process working either at pH 6.8 or 5.0 with two recombinant laccases, one constructed from the *Myceliophthora thermophila* enzyme and expressed in *Aspergillus oryzae* (rMt), and the other from the *Pycnoporus cinnabarinus* enzyme, respectively, led to a similar and high production yield of pyomelanin (Lorquin et al., 2021). Most probably from BQA, enzymatic decarboxylation by the rMt laccase was highlighted and led to the partial formation of gentisyl alcohol (major) and gentisaldehyde (minor compound) identified by ¹³C solid-state NMR, which were integral parts of the polymer. These two decarboxylation-issued products represented 11–13% of the total components, in reverse order of level compared to the abiotic HGA decarboxylation described by Liebgott et al. (2008). Some remarks are needed here. The redox potential of HGA is 636 mV (Eslami et al., 2014), that of the rMt laccase 450–480 mV (Berka et al., 1997), but that of *P. cinnabarinus* laccase is 810 mV (Sigoillot et al., 2004), a value >636 mV, hence, a study that requires more investigations. The polymerization reaction by laccases does not require a mediator typically used for depolymerization; such a mediator also has no effect in the case of HGA polymerization by rMt laccase (Lorquin et al., 2021).

Polymerization Mechanism and Structure Data Assembly Mode

Despite the attempts of simulated 3D structures that were mapped with the Fukui function by Turick et al. (2010), FTIR, X-rays, mass spectrometry, and NMR techniques were never able to deliver consistent conclusions in the exact conformation and sequencing of pyomelanin, and self-aggregation studies were never completed. However, recent comparative studies have provided important insights. Whether for the synthetic (abiotic-alkaline), bacterial, or enzymatic (laccase) polymerization method, the high reactivity of the primary phenoxy radicals in favor of aryl radicals had been observed and confirmed the close analogy with the hydroquinone polymerization (Lorquin et al., 2021; Sun et al., 2013; Fig. 4A). The presence of C_{ar}–C_{ar} linkages only and the total absence of C_{ar}–O–C_{ar} (ether linkages) resonance in the ¹³C solid-state NMR spectra of the three polymers were also demonstrated. In the absence of suitable mass spectrometry and hydrolysis methods for exact monomer composition, the three structures were characterized by solid-state NMR and FTIR. The mechanism in Fig. 4A proposes the most and less probable dimer assemblies drawn from aryl radical generation and discriminated from their steric hindrances. Ultimately, from the C₄–C₆ (α-bindings) and C₃–C₆ (β-bindings) attachments on the HGA monomer, the eight resulting dimer configurations represented in Fig. 4B are the most advantageous arrangements less subject to steric effects and confirm the previous observations on hydroquinone polymerization (Sun et al., 2013). These eight structures let us imagine the multitude of possible combinations with these dimers forming the pyomelanin, which cannot be technically differentiated at this time. To summarize, the polymerization of HGA follows the same mechanism in the three types of procedure except that (i) gentisyl alcohol and gentisaldehyde are integral parts of the bacterial (Py_OBACT) and laccase (Py_OENZ) polymers due to the decarboxylation by laccases (Lorquin et al., 2021), and (ii) the laccase yield is much higher than that of the chemical autoxidation. From the large dispersity value reported for the polymers, mainly Py_OENZ and Py_OBACT (see next section), self-aggregation modes (π–π stacking, ionic bonds), and building blocks involved need to be elucidated.

Table 1. Summary of the Highest Pymelanin Production Methods. *Value Recalculated from that Corresponding to bBioreactors Containing 7 L Leaches and 2 kg of Black Soldier Fly Larvae (See Text). 2,5-DMPA, 2,5-Dimethoxyphenylacetic Acid; YE, Yeast Extract; YPD, YE-Peptone-Dextrose; And YNB, Yeast Nitrogen Base

Microorganism or process	Expressed gene, mutagenesis type	Melanin yield	Phenolic precursor	C-N source	T °C	Reference
Pyomelanin HGA-Mn ²⁺ autooxidation (optimized process)	-	0.317 g g ⁻¹	2,5-DMPA	-	30	Lorquin et al. (2021)
4-HPPD recombinant enzyme (from <i>Pseudomonas aeruginosa</i> PAO1 st.)	Heterologous expression, heat shock transformation	0.213 g L ⁻¹	L-Tyr	Glucose, casamino acids	37	Bolognese et al. (2019)
4-HPPD recombinant enzyme (from <i>Ralstonia pickettii</i>)	Heterologous expression in <i>E. coli</i> KSYH	0.315 g L ⁻¹	L-Tyr	LB medium with Cu ²⁺	30	Seo & Choi (2020)
<i>Pseudomonas putida</i> F6 mutant (F6-HIDO)	Random mutagenesis (transposon Tn5)	0.35 g L ⁻¹	L-Tyr	Na ₃ -citrate, dextrose	30	Nicodinovic-Runic et al. (2009)
<i>Halomonas titanicae</i> (wild strain)	-	0.55 g L ⁻¹	4-HPA	Basal medium with YE	30	Lorquin et al. (2021)
<i>Yarrowia lipolytica</i> W29 (wild strain)	-	0.50 g L ⁻¹	L-Tyr	YNB medium with glucose, Asp, and Gly	30	Ben Tahar et al. (2019a)
Laccase extract polymerization (on-line optimized process)	Recombinant laccase from <i>M. thermophile</i> (fMt laccase)	1.25 g g ⁻¹	2,5-DMPA	-	30	Lorquin et al. (2021)
4-HPPD recombinant enzyme (from <i>Yarrowia lipolytica</i> W29)	Overexpressed gene (3 copies) by the JMY8032 strain construct	4.50 g L ⁻¹	L-Phe	YPD medium with glucose, Leu, uracil	37	Larroude et al. (2021b), See also construction in 2021a)
Pyomelanin-eumelanin (mixt <i>Clostridium, Lactobacillus</i> , and flies (2 bioreactors process, see text))	-	5.35 g L ^{-1*}	-	Leachates, nutrient-rich solution	20	Popa & Nealson (2014)

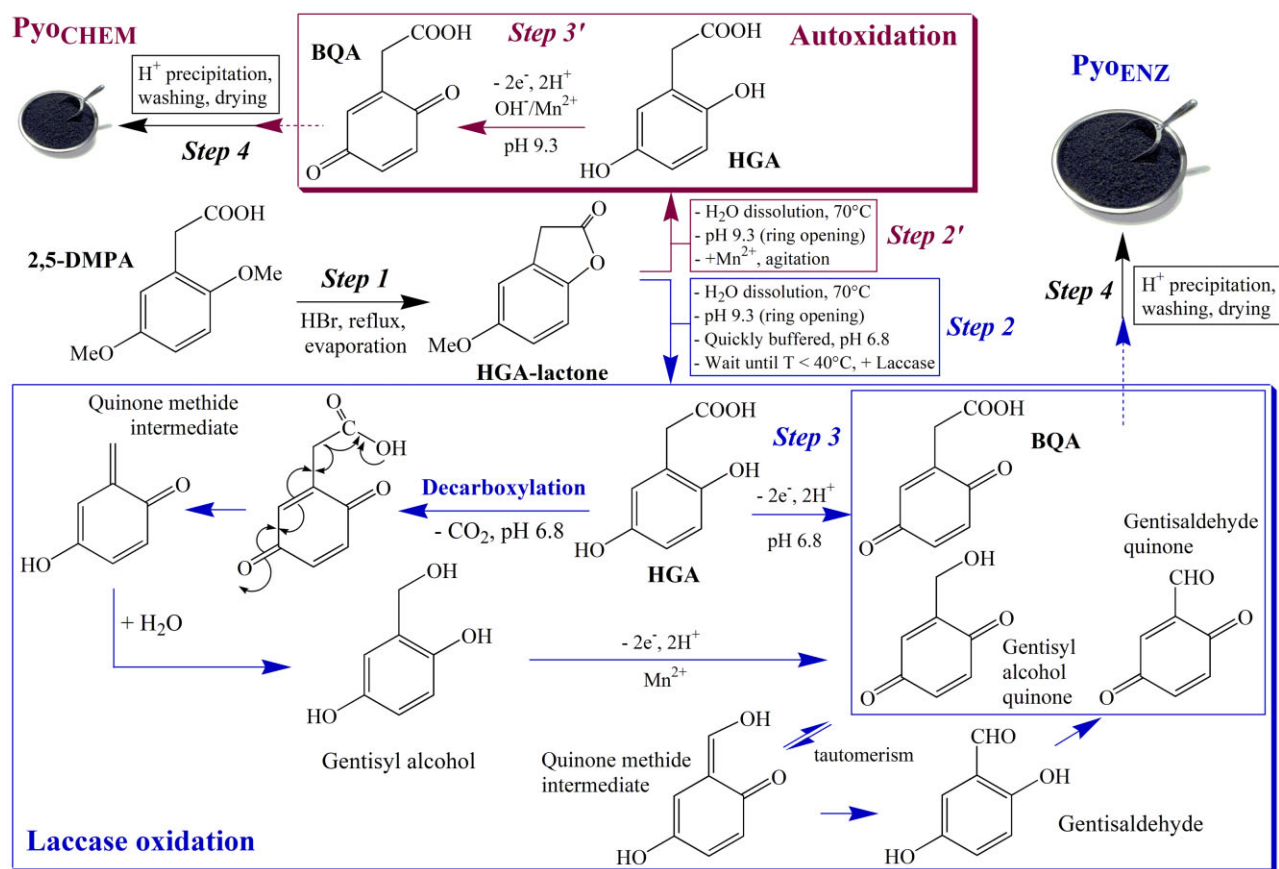


Fig. 3 Schematic diagram of the production of pyomelanin by the laccase process (in blue) (reproduced with permission of Lorquin et al., 2021). Comparison to the abiotic autoxidation process (in red) via steps 2' and 3'. The laccase oxidation generates a decarboxylation mechanism giving gentsyl alcohol and gentsaldehyde identified in the polymer. In step 1, 2,5-DMPA is demethylated by HBr at reflux and gives HGA-lactone at 99% yield. In step 2, the lactone is opened and the laccase is added at its optimal activity pH. In the case of the rMt laccase, the reaction medium is previously buffered at pH 6.8 (optimal activity). Step 4 is the final HCl precipitation, followed by washing and drying. Compounds: BQA, 1,4-benzoquinone acetic acid; gentsaldehyde, 2,5-dihydroxybenzaldehyde; gentsyl alcohol, 2,5-dihydroxybenzyl alcohol; 2,5-DMPA, 2,5-dimethoxyphenylacetic acid; and HGA, homogentisic acid.

Molecular Weight and Spectroscopic Features

The size of pyomelanin was first evaluated from the pigment of the bacterium *A. eutrophus*, the molecular weight (M_w) of 3000 Da estimated by GPC/SEC (gel permeation/size exclusion chromatography), however using unconventional polyethylene glycol as standard, while the synthetic pyomelanin (autoxidized HGA) was found at a smaller size of 1700 Da (David et al., 1996). In the absence of a suitable mass spectrometry technique, the GPC/SEC method on an adapted aqueous column remains the most appropriate, provided that an alkaline eluent ensures the solubility of the pigment to avoid high M_w agglomerates and the calibration line performed with similar polymers such as poly(styrene sulfonates) strongly recommended. Finally, the three optimized processes led to close M_w of 5400 (dispersity D 15.3) for the enzymatic (Pyo_{ENZ}; general formula C₂₂₁H₁₄₀N₁₁O₁₅₁), 5700 Da (D 11.9) for the bacterial (Pyo_{BACT}; C₁₉₃H₁₇₉N₁₅O₁₈₇), and a less high M_w of 2300 Da (D 6.64) for the autoxidized pyomelanin (Pyo_{CHEM}; C₁₀₅H₅₉O₆₁) (Lorquin et al., 2021). These sizes are higher than those of laccase-synthesized polymers of catechol (M_w 1268 Da), resorcinol (1489 Da), and hydroquinone (1157 Da) (Sun et al., 2013). Melanins have also paramagnetic characters, EPR (or ESR; electron paramagnetic [spin] resonance) techniques were successful to determine the free radical centers that absorb microwaves

under a magnetic field, thereby generating a characteristic signal at 324–325 mT, for instance for pyomelanin of *C. neoformans* and *Rubrivivax benzoatilyticus* (Frasés et al., 2007; Mekala et al., 2019). The UV-visible spectrum of pyomelanin is very wide (200–700 nm), the molar extinction coefficient is low, like that of eumelanin, from 7500 to 265 cm⁻¹ mol⁻¹ L for 200 to 820 nm, respectively (Sarna & Swartz, 2006). Sometimes, pyomelanin UV spectra could show an absorption in the 200–300 nm region that can be attributed to more complex structures and $\pi \rightarrow \pi^*$ and $n \rightarrow \pi^*$ transitions of amino, carboxylic, and aromatic moieties. Much less informative, other spectroscopic methods such as X-ray diffraction generate broad diffraction features due to the absence of regularly repeating units in polymers (Mekala et al., 2019).

Properties of Pyomelanin

Despite the need to the extent its properties, pyomelanin has been distinguished from those of the largely available cuttlefish ink bag of *Sepia officinalis* (well-characterized by Magarelli et al., 2010) and synthetic Dopa-melanin (see preparation in d'Ischia et al., 2013), giving advantages to pyomelanin (Lorquin et al., 2021).

Antimicrobial Activity

Pyomelanin is not essential for the growth of microorganisms but provides some advantages to their producers to cope with

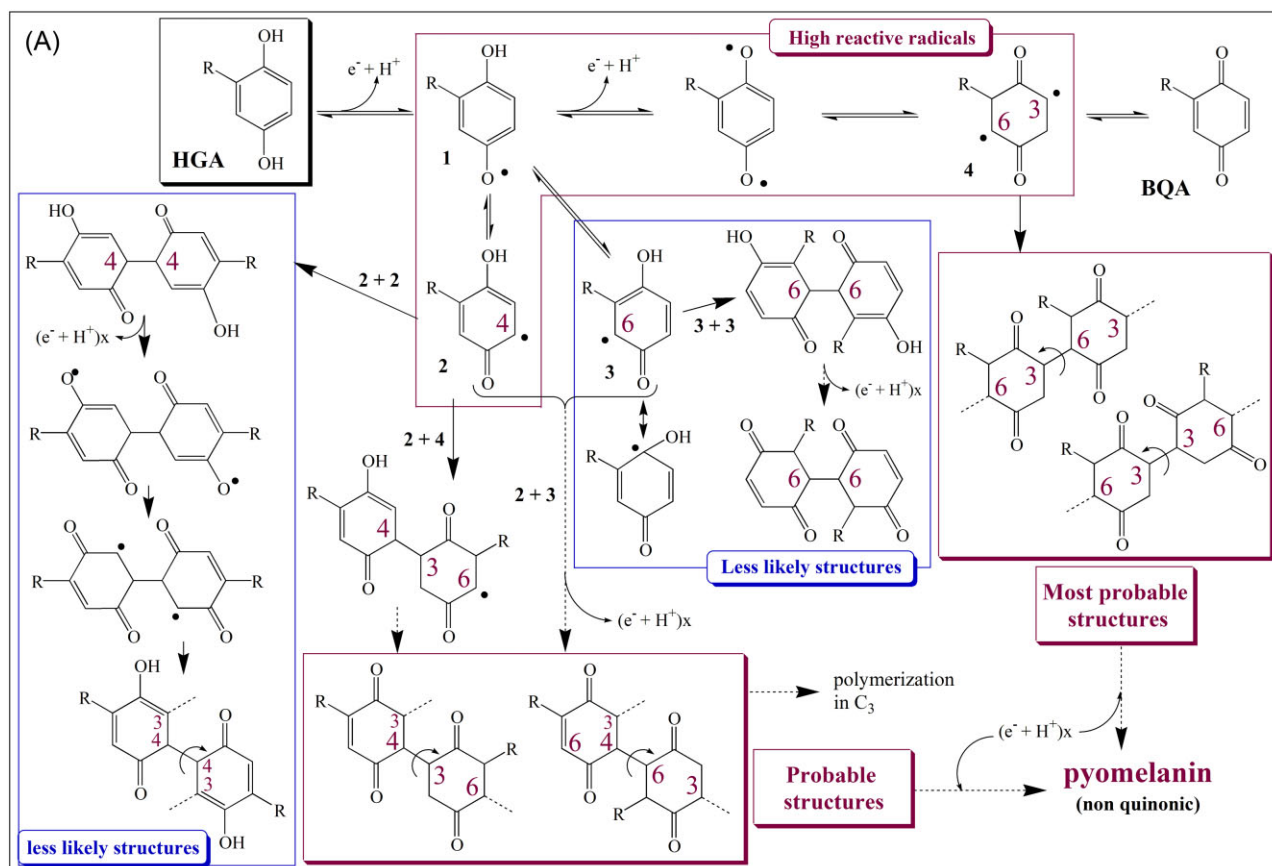


Fig. 4 The proposed mechanism for the radical polymerization of HGA, showing the most probable structures (A, in red) and the detailed representations of the corresponding dimers in (B). R = -CH₂-COOH. Gentsyl alcohol (major) and gentsaldehyde (minor) issued from the decarboxylation mechanism (laccase process, bacteria cultures) are not represented here but are incorporated into the polymer in the same manner as HGA radicals at locations of the chain that could not be determined at this time.

adverse challenges. *In vivo*, pyomelanin contributes to antimicrobial resistance and microbial pathogenesis as it is associated with virulence in a broad range of pathogenic fungi and bacteria (Nosanchuck & Casadevall, 2003). Notably, it was demonstrated that the HGO enzyme inactivation of *P. aeruginosa* and *Burkholderia cepacia* and the consequent pyomelanin accumulation led to a better adaptation to chronic infection in patients (Rodríguez-Rojas et al., 2009; Zughaier et al., 1999). The antibacterial and antifungal properties of the pigment purified from *P. aeruginosa* have been reported (Abdul-Hussien & Atia, 2017). However, the experiments were conducted on a few strains, which require more extensive investigations. The pyomelanin virulence also encompasses its action against antibiotics on which it binds, leading to the decrease of their antibacterial activities (Almeida-Paes et al., 2012, 2016; Liaw et al., 2010).

UV-Antiradical Properties and (Photo)Stability

The main property of pyomelanin is to protect microorganisms from UV light limiting free radical (ROS) generation, increasing resistance to light, for instance in *Legionella* (Steinert et al., 1995). Generally, UVA induces damage by directly transferring energy or indirectly through ROS generated as primary and secondary radiolytic products (Maccarrone et al., 1997). The antiradical role of pyomelanin *in vivo* has been demonstrated in *P. aeruginosa* (Boles & Singh, 2008), *B. cepacia* (Keith et al., 2007), *A. fumigatus* (Schmaler-Ripcke et al., 2009), and many other strains. Pyomelanin has never been described as a producer of ROS under UV irradiation. The

laccase-prepared pigment effectively scavenged ROS generated by UVA-visible irradiated keratinocyte with an IC₅₀ of 82.2 μg mL⁻¹, while the IC₅₀ of synthetic melanin (Mel_{SYNTH}, 284.1 μg mL⁻¹) was higher than that of *S. officinalis* (Mel_{SEPIA}) eumelanin very far (Lorquin et al., 2021). Comparatively, excitation of human eumelanin/pheomelanin by UV light produces ROS (singlet oxygen, superoxide, peroxide, and the hydroxyl radical) meanwhile they photoconsume O₂ and are protective against skin cancer (Szewczyk et al., 2016). Human melanin transforms UV radiation (optical energy) into heat, a mechanism still not well understood with the sole link to the skin burning for lack of melanin. To this day, we do not know whether pyomelanin has this ability. If the human melanin particle size is one of the contributing factors to the variations in skin cancer rates among skin types (Nofsinger et al., 1999), size-dependent photoreactivity studies are needed for pyomelanin. Like human melanin, the HGA pigment displayed an electrochemical response to UV light when combined with a carbon paste electrode and reacted by offering protection from the free-radical formation (Turick et al., 2010). Human eumelanin and pheomelanin photochemically generate degradation products that are responsible for sunlight-induced melanoma formation by inducing cyclobutane-pyridine dimers from DNA (Premi et al., 2015). In contrast, strong UVA-visible irradiation at 550 W m⁻² energy during 1 hr (200 J cm⁻²) on purified pyomelanin in solution, a dose much higher than the minima (72 J cm⁻²) imposed by the European Medicines Agency for photostability studies, did not generate degradation (Lorquin et al., 2021). Hence, pyomelanin

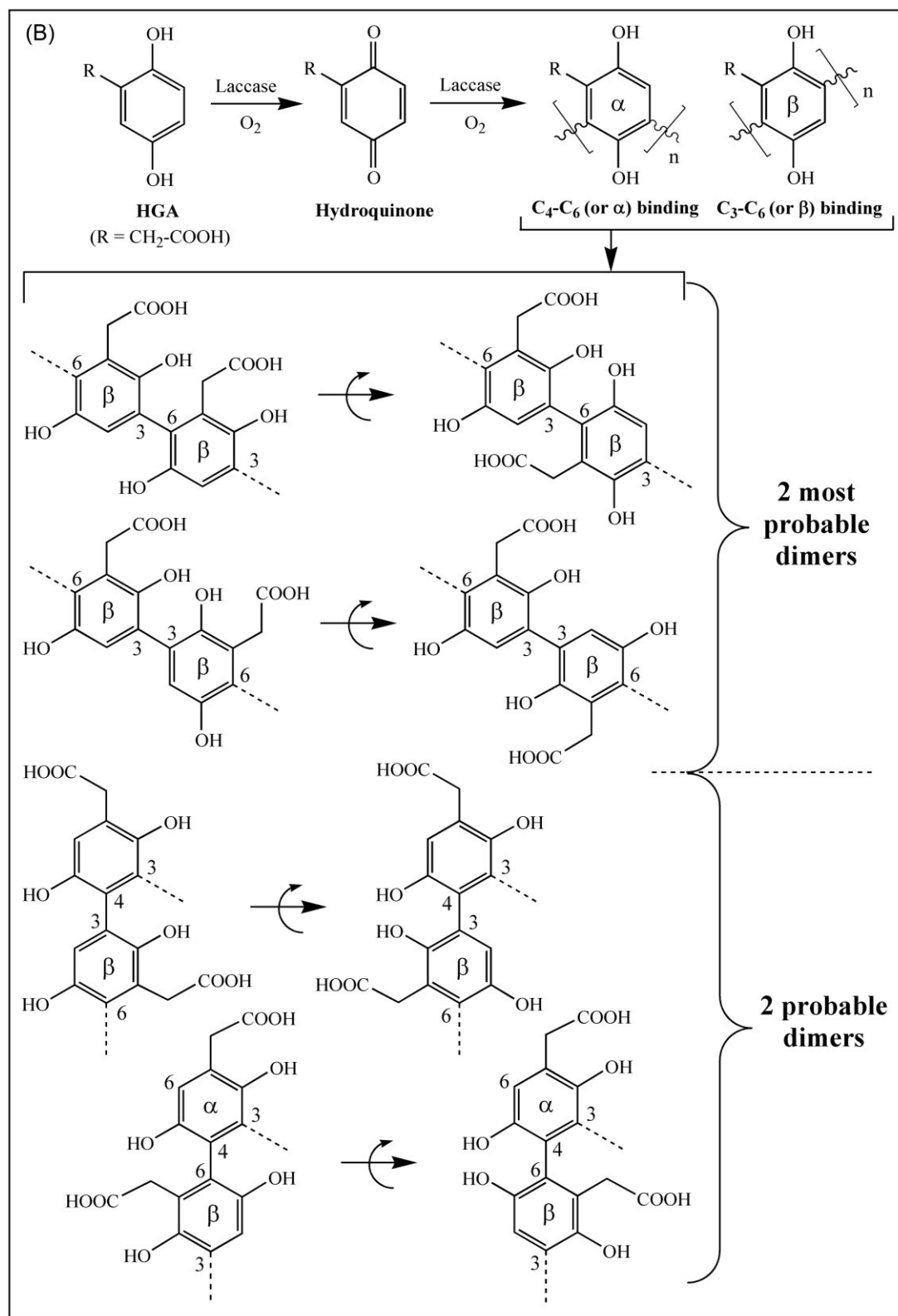


Fig. 4 Continued

is not a photosensitizing polymer, but is highly photostable and hyperthermostable (tested at 80°C for several days) due to the C_{ar} - C_{ar} linkages constituting the polymer assembly.

Against Oxidative Stress

The pigment diminishes the oxidizing stress of the host microorganism by its high tolerance to H_2O_2 and by a specific control due to several regulatory factors, as demonstrated in *Ralstonia solanacearum* (Ahmad et al., 2016). These findings reinforced previous results on oxidative stress from peroxide that decreased in clinical isolates of *P. aeruginosa* from infected patients because of pyomelanin synthesis (Rodríguez-Rojas et al., 2009).

DPPH-Antioxidant Activity

The *in vitro* conventional DPPH (1,1-diphenyl-2-picrylhydrazyl) assay was rarely reported due to its insolubility in organic solvents, and because the stable DPPH reagent reacts at slightly alkaline pH values. Despite this constraint, the purified pyomelanin from *Y. lipolytica* strain W29 at a low concentration in methanol was shown to exhibit a weak radical scavenging activity (EC_{50} 230 $\mu\text{g mL}^{-1}$; Ben Tahar et al., 2019b). Further assays on the three types of HGA pigments prepared in DMSO along with two common L-Dopa melanins (Mel_{SEPIA} , Mel_{SYNTH}) were successful. They indicated that the laccase process-issued pyomelanin (Pyo_{ENZ} , EC_{50} 27.5 $\mu\text{g mL}^{-1}$) and the synthetic pigment (H_2O_2 -oxidized L-Dopa, Mel_{SYNTH} , EC_{50} 25.9 $\mu\text{g mL}^{-1}$) have a DPPH-antioxidant activity equivalent to that of ascorbic acid (29 $\mu\text{g mL}^{-1}$) (Lorquin et al., 2021), similarly to pyomelanin from *P. stutzeri* strain BTCZ10 and *Pseudoalteromonas lipolytica* BTCZ28 cultures (Kurian & Bhat, 2018; Narayanan et al., 2019). Barely better, the EC_{50} of pyomelanin optimally obtained from autoxidized HGA (Pyo_{CHEM}) was 20.0 $\mu\text{g mL}^{-1}$. These EC_{50} values are far from that of pyomelanin of *H. titanicae* (130.0 $\mu\text{g mL}^{-1}$) and the abundant eumelanin of *S. officinalis* (Mel_{SEPIA} , IC_{50} > 300 $\mu\text{g mL}^{-1}$) (Lorquin et al., 2021).

Electron Transfer and Fe^{3+} -Reducing Activities

Due to its redox cycling nature, *in vivo* as well as *in vitro*, pyomelanin was demonstrated to serve as a terminal electron acceptor, electron shuttle, or conduit for electrons (Turick et al., 2008a, 2010; see applications below). Consequently, pyomelanin increases the current response of biofilms for electricity production in microbial fuel cells. Moreover, microbial melanin synthesis might be a mechanism to mobilize cations from the ecosystem and store them. All melanins in the reduced form can anaerobically reduce ferric to ferrous oxide. Pyomelanin was shown to be able to transfer electrons, in the reduction of soluble FeIII in FeII, and especially to ensure homeostasis and an appropriate Fe^{2+}/Fe^{3+} ratio for their survival, particularly when transporters and siderophores are absent (Andrews et al., 2003; Turick et al., 2002, 2003). In *L. pneumophila*, both HGA and pyomelanin mediate Fe^{3+} reduction, the resulting Fe^{2+} being available to the bacterium for uptake (Zheng et al., 2013). Under low dissolved oxygen levels, the HGA pigment accelerates the solid-phase metal reduction and aids in the survival of *Shewanella oneidensis* MR-1 (Turick et al., 2009). By an adapted ferrozine assay, the laccase-processed polymer (Pyo_{ENZ}), HGA-autoxidized pigment (Pyo_{CHEM}), and synthetic L-Dopa melanin (Mel_{SYNTH}) exhibited a similar and highest Fe^{3+} -reducing activity evaluated for Pyo_{ENZ} at 5.30 $\text{ng Fe}^{2+} \text{hr}^{-1} \mu\text{g}^{-1}$ pyomelanin (Lorquin et al., 2021).

A Moderate Anti-Inflammatory Activity

Generally, minimizing ROS minimizes inflammation in mammalian cells. By the LPS-activated murine macrophage RAW 264.7 cells to promote $NO\cdot$ radicals as the anti-inflammatory assays, the only recorded result is that of the pyomelanin isolated from the *P. lipolytica* BTCZ28 strain and conditioned under ultra-small pyomelanin nanogranules (PNG; Narayanan et al., 2019). The pigment showed a moderate $NO\cdot$ reduction of about 30% for 100 $\mu\text{g mL}^{-1}$ of PNG, and in a complementary way, inhibition of cyclooxygenase (IC_{50} 95.5 $\mu\text{g mL}^{-1}$), lipoxygenase (IC_{50} 88.9 $\mu\text{g mL}^{-1}$), and myeloperoxidase (fivefold diminution of the activity at 100 $\mu\text{g mL}^{-1}$) were also reported from the cell lysate of this strain, giving a proof-of-principle of using pyomelanin to diminish inflammation in therapeutic applications.

Pyomelanin Production

Previous sections have highlighted that the production of pyomelanin by abiotic HGA autoxidation could not exceed 0.317 g g^{-1} 2,5-DMPA (Table 1), and as formerly suggested, seems unable to compete with microbial culture processes (David et al., 1996; Ruzafa et al., 1995). That might be explained in part by ^{13}C solid-state NMR experiments that revealed a loss (~40%) of carboxylic moiety without observable by-product degradation during the alkaline autoxidation of HGA- Mn^{2+} (Lorquin et al., 2021). With the absence of an efficient coordination catalyst able to prevent this loss and the feasibility of microbial and enzyme-catalyzed pyomelanin syntheses, several strategies may be proposed.

Processes Using Wild Type Cells

Despite the great number of published pyomelanin-producing microorganisms, few works quantified HGA-melanin. Faced with the small quantities furnished, Turick et al. (2010) estimated they may produce between 1 and 115 femtograms of pyomelanin per cell, a 1–10 fg yield seems more common, which would approximately correspond to 0.03–0.3% pyomelanin relative to dry weight. A microbial development might logically use L-Tyr or 4-HPA supplemented wild-cultures, the absence of iron may prevent HGO and 4-HPPD activities while promoting that of the 4-HPAH-1 when present, and vice versa. Furthermore, instead of using organic HGO inhibitors that might polymerize with pyomelanin and affect its properties, other ions than iron should be selected to inhibit the HGO enzyme (see Table in Supplementary Material). Precautions must also be taken when using ascorbic acid or other reducing agents that activate or inhibit the HGO activity (Adachi et al., 1966; Schmidt et al., 1995; Veldhuizen et al., 2005). A pH-dependent reduction or oxidation of the iron atom bound to the enzyme site has been suggested but not yet well controlled. Among the genus *Halomonas*, HGA producers are easily cultivable strains at low NaCl concentrations in the presence of reduced N and C sources, and overproduce pyomelanin from 4-HPA by preculture inductions in the absence of Fe^{2+} (Liebgott et al., 2008). An induced and optimized culture of *H. titanicae*, a γ -proteobacterium isolated on the Titanic wreck, was also shown to synthesize the pigment through the 4-HPAH-1 bypass, however with a great amount of 4-HPA, at a yield equivalent to that of the yeast *Y. lipolytica* cultivated with L-Tyr (Table 1). At least an original patented process developed by Popa & Neelson (2014) furnishes a mixture of pyomelanin and eumelanin (% composition not given) from two controlled primary and secondary reactors in series, by repetitive trophic cycling with fermentation leachates or other low-cost nutrient-rich solutions. Nutrients are cycled

between bacteria of the genera *Clostridium* and *Lactobacillus*, and black soldier fly larvae, *Hermetia illucens*, where polysaccharides are transformed into natural melanins (detailed in Supplementary Material). The patent reports that bioreactors containing 7 L of primary leachate, 2 kg of larvae, and incubated for 12–14 days at 20°C, produce a 30–37.5 g melanin mixture (Table 1).

Comparatively, the two main melanins are produced in much larger quantities. From L-Tyr, L-Dopa melanin yielded 27.9 g L⁻¹ by the basidiomycete *Armillaria cepistipes* (Ribera et al., 2019). The production of DHN-melanin from glucose reached the highest melanin yield at 50 g L⁻¹ by the *Vibrio alginolyticus* MMRF534 strain (Vijayan et al., 2017).

Engineered Cells

The random mutagenesis (transposon Tn5) on the *P. putida* strain F6 generated mutants from which one has the gene encoding HGO disrupted. However, random mutagenesis on unknown sites is limited by the rational strategies used for further strain improvement as well as the phenotypic instability. As an important tool but never exploited for production, mutagenesis of the HGO enzyme (*ΔhmgA*) in iron-supplemented media has been demonstrated in *A. nidulans* (Fernández-Cañón & Peñalva, 1995), *A. fumigatus* (Schmaler-Ripcke et al., 2009), and *Bacillus thuringiensis* (Tan et al., 2019). Overexpression of the 4-HPPD gene in an *E. coli* strain cultivated on a simple C source (glucose, glycerol, and L-Tyr) remains the appropriate strategy (reviewed in Martínez et al., 2019). *E. coli* possesses endogenous aromatic amino acid transaminase, TyrB enzyme, and is HGO (-). A cosmid library had been generated from the marine bacterium *S. colwelliana* known to produce L-Dopa in addition to HGA, into *E. coli* from which one transformant produced melanin, the DNA sequences allowed the identification of the 4-HPPD (Fuqua et al., 1991). Many recombinant 4-HPPDs have further been developed in *E. coli*, such as those from *P. fluorescens* (Borowski et al., 2004; Wójcik et al., 2014), plants such as *Arabidopsis thaliana* and carrot (Lederer & Böger, 2005; Raspail et al., 2011), and human cDNA constructs to find effective inhibitors against ALK (Aarenstrup et al., 2002; Neuckermans et al., 2019). From these publications, no pyomelanin yield had been reported until the recent heterologous expression of the 4-HPPD gene from *Ralstonia pickettii* (Seo & Choi, 2020; Table 1). Much more effective, the gene 4-HPPD of the yeast *Y. lipolytica* was overexpressed in the chassis strain with up to three copies, leading to the highest yield of pyomelanin (4.5 g L⁻¹; Table 1) and revealing that the yeast could convert L-Phe to L-Tyr or that the HGA pathway is strongly induced by L-Phe (Larroude et al., 2021b). Comparatively and from L-Tyr, note that an engineered *melC* (tyrosinase gene) transformed in *Streptomyces kathirae* produced L-Dopa melanin at a yield of 28.8 g L⁻¹ (Guo et al., 2015).

Let us now consider the bypassing route 2b through the 4-HPAH-1 enzyme (Fig. 1). This route exists in microorganisms, while the 4-HPPD path appears unique in plants, where 4-HPA is rarely identified, it might be a growth regulator. So far, no 4-HPAH-1 mutant has been developed. Gene sequence and cloning of the *D. acidovorans* enzyme had been given for the first time by Zink et al. in 2008 (Bayer's group patent, updated in 2012). The protein showed two components. The first one is *hpaH*, which codes for a flavo-protein NADH oxidase-dependent that transforms 4-HPA into a non-identified metabolite called Z; and the second is *hpaC* (HPAC), a peptide also designed as a 1,4-dihydroxyphenylacetate mutase (EC 5.4.99.-; Raspail et al., 2011) catalyzes the conversion of Z into HGA. To make plants tolerant to herbicide treatments, the strategy of Bayer's group was to implement the pathway of metabolic

bypassing of 4-HPPD, by cloning in the plant the 4-HPAH-1 enzyme through *hpaH* and *hpaC* from *D. acidovorans* (Hareland et al., 1975) together with the 4-HPPO from *Arthrobacter globiformis* (Blakley, 1977). However, no HGA formation has been found from these constructs. In our laboratory, *hpaH* and *hpaC* genes from *Ruegeria pomeroyi* DDS3 and *H. olivaria* strains were cloned in *E. coli* BL21 (unpublished). Unfortunately, no HGA had been detected from the transformant cultures containing 4-HPA, suggesting that another DNA sequence might be involved in restoring the hydroxylase activity. Cloning an active 4-HPAH-1 would be further investigated regarding the low-cost 4-HPA substrate that would directly be converted to HGA with less energy consumption.

Process Using Laccase

Laccases are responsible for the oxidation-polymerization of HGA (see previous sections). Hence, a simpler enzymatic process can replace microbial cultures, avoiding the constraints of sterility and control, allowing higher temperatures (laccases may work at 50°C), and is the ideal method for products that require predictable quality and redox properties. The current price of HGA proposed by suppliers in the world has become overestimated, about €800 per gram. Generally, the chemical syntheses of HGA are not applicable for a large-scale process because of the high cost of the starting reagents, the number of steps, and the final yield between 50 and 70% when indicated (Bloomer & Damodaran, 1980; Bostock & Renfrew, 1978; Leaf & Neuberger, 1948; Prasad & Adapa, 1992). Finally, the former procedure of DeForrest Abbott & Doyle-Smyth (1949), starting from 2,5-DMPA and through HGA-lactone, remains the most convenient, in a three-step online process, which can be widely large-scale extrapolated (Fig. 3). When using the concentrated extract of the rMt laccase, a high pyomelanin yield was obtained (Py_{ENZ}; Table 1). The 2,5-DMPA could as well be easily prepared from 2,5-dimethoxyacetophenone, four times less expensive, by a Willgerodt-Kindler reaction.

Applications: Toward Realistic Uses General Considerations

The degradation of pyomelanin by enzymes or microorganisms has never been reported to date. The main brake would be the pharmacological administration of the pigment further in contact with connective tissues. In this respect, it has been suggested that mono and small oligomers of BQA associated with oxidative stress would define the "ochronotic pigment" in ALK (Arnoux et al., 2015). What would happen if higher-sized pyomelanin such as Py_{ENZ} is administered, while in this case, there is no generation of free radicals and no arthritis by analogy to ALK. Waiting for investigations on the impact of high-size polymers on connective tissues, health applications are thus limited to cosmetology. On the other hand, is the pigment solubility an issue? Pyomelanin is not a soluble pigment; very fine particles are generally in suspension in neutral aqueous and most organic media. While waiting to find a pharmaceutical solvent, the only alternative remains to dissolve the dried pigment at 10 mg mL⁻¹ in 0.05 N NaOH, or in special cases at 0.5 mg mL⁻¹ in DMSO, which are the max. dissolved concentrations for preparing stock solutions (Lorquin et al., 2021). Moreover, the addition or presence of polyethylene glycol, proteins, lipids, and amines or amino acids just before the HGA polymerization might improve pigment solubility (Aravindakshan & Haberman, 1998; Stoner & Blivaiss, 1967). Note that melanin derived from natural sources, and most likely pyomelanin, generally shows chemical versatility by conjugation with other molecules, however the

advantage of good biocompatibility with no antigenic response in living cells (Araújo et al., 2014; Bettinger et al., 2009).

Applications in Cosmetology: Toxicological Aspects and Perspectives

Cytotoxicity is generally evaluated through two standard and complementary procedures (cell viability and acridine orange/ethidium bromide coloration) toward two mammalian cells, namely, mouse fibroblasts (NIH3T3) and human keratinocytes (HaCaT), that are used as a part of the 3D skin model for the assessment of the toxic hazard of cosmetic ingredients (Pfuhrer et al., 2014). By these methods, pyomelanin extracted from the yeast *Y. lipolytica* was found as non-cytotoxic at 100 $\mu\text{g mL}^{-1}$ (Ben Tahar et al., 2019b), and the same pigment-based gold nanoparticles showed neither cytotoxicity effect nor altered cell morphology (Ben Tahar et al., 2019a). Narayanan et al. (2019) also showed that pyomelanin from *P. lipolytica* strain BTCZ28 was a non-toxic pigment, which exhibits only ~25% cell death at 100 $\mu\text{g mL}^{-1}$, on fibroblast L929 cell line and by using the MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) assay. Cytotoxicity toward human keratinocytes of Pyo_{ENZ} (laccase), Pyo_{BACT} (bacterial), and Pyo_{CHEM} (autoxidized HGA process), evaluated by the vital dye neutral red penetration technique, was found non-cytotoxic until 500 $\mu\text{g mL}^{-1}$ (Lorquin et al., 2021). Studies have also been conducted for the action of pyomelanin on cancer cells *in vitro*. Purified pyomelanin from cultures of *P. aeruginosa* strains isolated from patients was assessed on lung A549 (IC₅₀ 410.1 $\mu\text{g mL}^{-1}$), skin A375 (234.4 $\mu\text{g mL}^{-1}$), macrophage RAW264.7 (235.4 $\mu\text{g mL}^{-1}$) cancer cell lines, and Vero (kidney, 100 $\mu\text{g mL}^{-1}$) cell line, and showed moderate cytotoxicity with a dose-dependent effect (Mahmood et al., 2016). Much better, the tumor cell kill-effect of pyomelanin from the strain *P. lipolytica* BTCZ28 was estimated in the lung carcinoma cell line A549 with an IC₅₀ value at 96.1 $\mu\text{g mL}^{-1}$ (Narayanan et al., 2019). In summary, pyomelanin appeared non-cytotoxic against fibroblast and keratinocyte, and cytotoxic against cancer cell lines two characteristics of importance for skin health.

While two cases of melanin are used in cosmetics, a catechol-melanin from black olive fruits against sunburns and aging (Laboratoires Biocyte, France), and a non-communicated biomelanin in sunscreen lotions (Baar Products, PA, USA), to date, pyomelanin has never been used in emulsion or other formulation types. Yet the pigment had been proposed for a long time as a potential ingredient in cosmetics and nanocosmetics for which we must address the aesthetics, conservation of the formulation, and its actual effect on the skin. While the brown color might seem unattractive, in emulsions the diluted pigment exerts a wood color and a natural aspect. It might also impart a natural tanned appearance when applied concentrated. The anti-inflammatory property of pyomelanin against skin disorders, even if moderate (see properties), is an excellent pharmacological complement. The pigment might treat post-inflammatory hypopigmentation as well, for instance in the cases of eczema, acne, trauma, burns, and psoriasis, as already suggested for mixtures of synthetic L-Dopa-melanin (Pawelek et al., 1995). In addition, due to its poor solubility notably, at pH 6, which is the pH of the skin and emulsions in general, there is a very high probability that high size pyomelanin does not permeate the stratum corneum barrier and reach the epidermis and dermis. Pyomelanin could also be formulated as an antimicrobial agent, a 0.05–0.1% max. concentration seems the most appropriate, as assayed in a sunscreen preparation containing a bacterial pigment (Kurian & Bhat, 2018). Extensive antibacte-

rial assays related to skin infections, such as acne and nosocomial bacteria, have become necessary.

Considering the action of pyomelanin against light radiation and contrary to what one might think, pyomelanin as well as L-Dopa melanin in no way is sunscreen because of the low sun protection factor (SPF) from 2 to 4 (Epstein et al., 2007). SPF must be especially evaluated with a spectrophotometer equipped with an integrating sphere accessory to collect the scattered light from poly(methyl-methacrylate) plates. However, pyomelanin was shown to slightly enhance the SPF value (Ben Tahar et al., 2019b; Kurian & Bhat, 2018). The pigment absorbs UV radiation notably UVA without being degraded, even by heat, and effectively scavenges ROS (see properties), consequently can protect the skin from premature aging (wrinkles) and the occurrence of cancers. In contrast, infrared rays (IRs) do not seem to cause cancer, but they contribute significantly to skin aging, especially those in the IRA range (770–1400 nm) and increase the concentration of metalloproteinase-1 (MMP-1) in dermal fibroblasts of the matrix (Krutmann et al., 2012). MMP-1 is a zinc-dependent protease that degrades collagen and other extracellular matrix molecules and is a potential target in cancer therapy, indicating the great need to protect against IR radiation for future formulations. Advantageously, MMP-1 mRNA expression was reduced significantly in the presence of antioxidants (Grether-Beck et al., 2015). Thus, due to its high antiradical activity, the incorporation of pyomelanin in a cosmetic formulation could be beneficial for the protection against UV as well as IR radiation.

As Material and Environmentally Friendly Protectant

Since pyomelanin absorbs light throughout the UV-visible spectra, it would also be effective as a glass or plastic tinting agent for eyeglasses, contact lenses, car and house windows, office buildings, etc. Likewise, such pigments would be effective agents in protecting industrial materials, the rate of deterioration of paint, wood, plastic, and rubber being dramatically increased by exposure to UV radiation. On the other hand, pyomelanin could serve as an antifouling agent. Indeed, it significantly reduces larval attachment and metamorphosis in the mussel *Mytilus coruscus*, hence preventing the attachment of fouling organisms particularly on boat hulls (Zeng et al., 2017). At least, pyomelanin chelates heavy metals, for instance, uranium present in some contaminated soils (Turick et al., 2008a, 2008b, 2010), as other melanin does.

As a Promising polymer for Bioelectronics and Energy

Melanins generally have some energy conversion ability, conductivity, and semiconductor properties. They collect energy from lower-energy radiation sources, and kick electrons into excited states, initiating a process that would end up producing chemical energy. Interestingly, the freestanding film of melanin electrochemically obtained from L-Dopa exhibited photoconductivity (Subianto et al., 2005). To explain this effect, it was demonstrated that (i) the increase in the electron-transfer properties of melanin is independent of the energy of the incident photons; (ii) melanin functions as a transducer, oxidizing water, pushing apart water molecules, as well as recruiting back ions into molecules that are subsequently polarized again; and (iii) melanin drives the photon energy of lower-energy radiation sources by quenching electrons and initiating an ionic event independently of their relative energy contention (Solis et al., 2007). The same mecha-

nism can be predicted for pyomelanin that exhibits an electrochemical response after UV-light exposition (see properties). The HGA pigment should thus be developed as a component for solar panel energy converters, with better advantages due to its high thermo- and photostability as well as highly protectant-material properties.

Pyomelanin has an important redox cycling nature with multiple quinone moieties, and the discharge or recharge of electrons from quinone centers is likely followed by the repartition of electrons and protons within the polymer until equilibrium is reached. Pyomelanin can thus conduct electricity like an electronic-ionic hybrid conductor for bioelectronic applications by only femtograms per cell (Turick et al., 2010), better than melanin. The HGA pigment has a great ability for electricity storage and could replace heavy metals in the construction of long-life, deep cycle, and rechargeable batteries of a new generation. Indeed, because most redox transformations involving melanin occur at low redox potential ($E < 1-2$ V), the variation in the energy level associated with electron exchanges is small relative to the strength of the covalent bonds, which hold the quinone structure together. Henceforth, pyomelanin should be incorporated into batteries that might be charged and discharged numerous times without affecting their stability. To replace the electron substrate with a heavy metal, a recent method for producing Li-pyomelanin (LPM), Na-pyomelanin (SPM), and K-pyomelanin (PPM) has been described as well as the related safety batteries (Popa et al., 2019). LPM, SPM, or PPM was placed in the negative compartment (anode), which led to the decreased tendencies to overheat and/or explode. Such battery developments deserve further investigations, especially using various molecular sizes of pyomelanin.

Conclusions

In this updated review, all microbial pathways, including regulation-inhibition of pyomelanin synthesis and structural data, have been extensively covered. These methodologies would be useful to find microorganisms or enzymes more efficient for the production of the pigment. At this time, we conclude the amount of available pyomelanin, the prime limiting factor for all these years, is no longer an obstacle, offering realistic and functional applications in highly coveted fields and ensuring the fair rehabilitation of the polymer. The production has been resolved by either using laccase and starting from 2,5-DMPA or the overexpressed *Y. lipolytica* 4-HPPD recombinant enzyme cultivated with L-Tyr, two very recent and convenient processes that can be redimensioned and still optimized for an industrial unit.

Supplementary Material

Supplementary material is available online at JIMB (www.academic.oup.com/jimb).

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Conflict of Interest

The authors declare that they do not have any conflicts of interest that prevent the writing of the present paper. This review was

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